# Unexpected O−H Insertion of Rhodium-Azavinylcarbenes with N‑Acylhydrazones: Divergent Synthesis of 3,6-Disubstituted- and 3,5,6-Trisubstituted-1,2,4-Triazines

Jiang Meng, Min Wen, Shiwei Zhang, Peiwen Pan, Xingxin Yu,\*<sup>1</sup> and Wei-Ping Deng\*

Shanghai Key Laboratory of New Drug Design & School of Pharmacy, East C[hina](#page-10-0) University of Science an[d T](#page-10-0)echnology, 130 Meilong Road, Shanghai 200237, China

**S** Supporting Information



ABSTRACT: A practical and efficient method for divergent synthesis of 3,6-disubstituted- and 3,5,6-trisubstituted-1,2,4-triazines via unexpected rhodium-catalyzed O−H insertion/rearrangement/conditions-controlled intramolecular cyclization and oxidation reaction under mild conditions has been developed. Notably, it is the first example for the synthesis of 1,2,4-triazines with different substituted-patterns via a common intermediate with excellent chemoselectivities by the reaction of N-acylhydrazones as aze-[3C] or  $[4C]$  synthons with N-sulfonyl-1,2,3-triazoles as aze- $[2C]$  synthons. Furthermore, this method allows direct access to di(het)aryl ketone frameworks containing 1,2,4-triazine moiety for the first time, serving as a versatile building block for the synthesis of other useful heterocyclic skeletons, such as pyridine or pyridazinone-fused triazine in excellent yields.

# ■ **INTRODUCTION**

Aromatic heterocycles are highly important structural units found in a large number of biologically active natural products, pharmaceutical compounds, and functional materials.<sup>1</sup> Among them, those nitrogen-containing heterocycles are the most indispensable structural motifs with great value i[n](#page-10-0) organic syntheses. 1,2,4-Triazines and its derivatives are a well-known class of azaheterocycles for their broad spectrum of biological activities and natural occurrences (Figure 1).<sup>2</sup> Taking Lamictal



for example, as a member of the sodium channel blocking class of antiepileptic drugs, it has been used in the treatment of epilepsy, bipolar disorder and so on for a long time. Apart from these impressive biological properties, 1,2,4-triazines have been widely used as 2-azadienes in the inverse electron demanded Diels−Alder reactions to build important heterocyclic skeletons.<sup>3</sup> Moreover, 1,2,4-triazines are also used as key metal coordinating ligands and fluorescent sensor.<sup>4</sup>

Given the wide usefulness of 1,2,4-triazines and their derivatives, considerable efforts have been devoted to the development of improved methods for their synthesis (Figure  $2$ ).<sup>5</sup> To date, several efficient methods are available for their construction. The mainstay for the synthesis of subs[tituted](#page-1-0) [1](#page-1-0),[2,4](#page-10-0)-triazine is the condensation of a 1,2-dicarbonyl equivalent or its precursor with an amidrazone.<sup>5</sup> However, many of these reported methods have drawbacks such as tedious substrate preparation procedure, harsh rea[ct](#page-10-0)ion conditions, limited substrates scope, and especially low regioselectivity. $5$  Therefore, the development of a more convenient and regiospecific methodology is highly desired.

Multiheteroatom heterocycles are of very considerable significance particularly in medicinal chemistry, for example analogues of the pyrimidine and purine nucleosides have been extensively studied.<sup>1</sup> In recent years, the rhodium-catalyzed denitrogenative transformations of readily accessible Nsulfonyl-1,2,3-triazo[le](#page-10-0)s have led to a new family of synthetic methods toward a variety of valuable organic molecules.<sup>6−11</sup> As shown, the in situ generated rhodium-azavinylcarbene (Rh-AVC) displays diverse reactivities, which could be [us](#page-10-0)[ed](#page-11-0) as  $[1C]$ -/ $[2C]$ -/aze- $[3C]$ -synthon (Figure 3a), especially for the synthesis of nitrogen-containing heterocycles. However, up to now, its potential as a aze- $[2C]$ -[synthon r](#page-1-0)emains untouched.<sup>9b</sup>

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Figure 2. Previous methods for the construction of 1,2,4-triazines.



electron-rich site

b) Previous Rh(II)-catalyzed N-H insertion



c) Initial design of formal Rh(II)-catalyzed [3+3] cycloaddition

Rh(II)-AVC





Also, there was, to the best of our knowledge, a rare method developed for synthesis of important heterocycles containing more than two heteroatoms in rhodium-azavinylcarbene chemistry owing to the challenge of choosing an appropriate

substrate.<sup>6−11</sup> With our research interests in azaheterocycle synthesis and N-sulfonyl-1,2,3-triazoles chemistry, $12$  we have reported [a](#page-10-0) [nov](#page-11-0)el and efficient route for the construction of oxabridged 2,5-epoxy-1,4-benzoxazepines via Rh(I[I\)-](#page-11-0)catalyzed reaction of N-sulfonyl-1,2,3-triazoles and salicylaldehydes. $12b$ We envisaged that the triazoles could be used to construct multiheteroatom heterocycles according to the high nucl[eo](#page-11-0)philicity feature of nitrogen atom in Rh-AVC (Figure 3a).

Formal 1,3-insertions of the Rh-AVC into N−H bonds of amides, ambiphilic β-enamino esters, 9H-carbazoles have been achieved (Figure 3b).<sup>13a-c</sup> Meanwhile, hydrazones have been widely used to react with olefins via  $[3 + 2]$  cycloaddition to form pyrazolidines. In[spi](#page-11-0)r[in](#page-11-0)gly, hydrazone may be the possible substrate for the construction of 1,2,4-triazine via a formal  $[3 +$ 3] cycloaddition owing to its feature containing two nitrogen atoms (Figure 3c). Herein, we describe a novel method for the divergent synthesis of 3,6-disubstituted- and 3,5,6-trisubstituted-1,2,4-triazines via an unexpected O−H insertion of Rh-AVC with N-acylhydrazones rather than the initial hypothesis (Figure 3d).

#### ■ RESULTS AND DISCUSSION

We initiated our studies by testing the reactions of different types of hydrazones with triazole 1a (Scheme 1). Disappointingly, the desired cycloaddition product was not detected, and there were even no reactions takin[g place in](#page-2-0) most cases. Surprisingly, when hydrazone 2a was used, an unexpected product 3aa was obtained in 57% yield. We reasoned that the hydroxy tautomer of benzoylhydrazone is easily formed under the reaction conditions and then could been captured by Rh-AVC through O−H insertion rather than the assumed N−H insertion.<sup>13c</sup> Interestingly, 3aa would degenerate after a long time exposing to air or remaining in NMR tube to form the expected [3,6](#page-11-0)-disubstituted-1,2,4-triazine 4aa. The encouraging results promoted us to do a further study of this transformation.

With the initial results in hand, we first focused on the improvement of rhodium-catalyzed reaction of triazole 1a with hydrazone 2a. Slightly increasing the amount of triazole 1a, the yield of 3aa was improved to 66% (Table 1, entry 2). Using 1,2-

<span id="page-2-0"></span>Table 1. Optimization of Reaction Conditions for the Synthesis of 3,6-Disubstituted-1,2,4-triazine  $4aa^a$ 

$N = N$ $N - Ts$	Ph				
Ph	1a	Rh(II), T <sub>1</sub> , t <sub>1</sub>	-Ts	additive	
	OEt	solvent Ph		$T_2, t_2$	Ph
Pŀ	2a	3aa			4aa
entry	solvent	Rh (II)	$T_1({}^{\circ}C)/t_1(h)$	3aa $(\%)^b$	4aa $(\%)^c$
1 <sup>d</sup>	CHCl <sub>3</sub>	$Rh_2(oct)_4$	75/3.5	57	
$\mathbf{2}$	CHCl <sub>3</sub>	$Rh_2(oct)_4$	75/3.5	66	
3	DCE	$Rh_2(oct)_4$	75/3.5	69	
$\overline{4}$	DCE	$Rh_2(oct)_4$	100/3.5	47	
5	DCE	$Rh_2(oct)_4$	65/3.5	68	
6	<b>DCE</b>	$Rh_2(oct)_4$	75/4	67	
7	DCE	$Rh_2(oct)_4$	75/3	69	
8	DCE	$Rh_2(piv)_4$	75/3	53	
9	<b>DCE</b>	$Rh_2(oAc)_4$	75/3	NP	
10	toluene	$Rh_2(oct)_4$	75/3	56	
11 <sup>e</sup>	<b>DCE</b>	$Rh_2(oct)_4$	75/1	69	
$12^e$	<b>DCE</b>	$Rh_2(oct)_4$	75/4	70	
$13^{e,f}$	<b>DCE</b>	$Rh_2(oct)_4$	75/1	75	
$14^{e,g}$	<b>DCE</b>	$Rh_2(oct)_4$	75/1	75	
$15^{e_i f_i h}$	<b>DCE</b>	$Rh_2(oct)_4$	75/1	75	68
$16^{e_i f,i}$	<b>DCE</b>	$Rh_2(oct)_4$	75/1	75	68
$17^{e,f,j}$	<b>DCE</b>	$Rh_2(oct)_4$	75/1	75	< 10
$18^{e\int k}$	<b>DCE</b>	$Rh_2(oct)_4$	75/1	75	trace
$19^{e,f,l}$	<b>DCE</b>	$Rh_2(oct)_4$	75/1	75	67

<sup>a</sup>Conditions: Under N<sub>2</sub>, 1a (0.24 mmol), 2a (0.2 mmol), Rh(II) (2 mol %), and solvent (1.0 mL) were heated at 75 °C until 2a was consumed, and then the next step was conducted. <sup>b</sup>Isolated yields. consumed, and then the next step was conducted. <sup>b</sup>Isolated yields.<br><sup>c</sup>Isolated yields based on intermediate 3. <sup>d</sup>1 equiv 1a was used. <sup>e</sup>5 mol % Rh(II) was used.  $f_1A$  equiv 1a was used.  $g_1$  S equiv 1a was used.  $h_2$ mol % p-TsOH was added, 2 mL THF was used as the solvent,  $T_2$  =  $30^\circ\text{C}$ ,  $t_2 = 4$  h. <sup>1</sup>The intermediate 3aa was directly exposed to air,  $T_2 =$  $30^\circ \text{C}, t_2 > 72 \text{ h}.$  120 mol % CF<sub>3</sub>COOH was used instead of p-TsOH.<br> $k_{20}^{k_{20}}$  or  $\frac{1}{20}$  mol % PhCOOH was used instead of n-TsOH  $^{16}$ CO mol % FeCl  $20 \text{ mol } \%$  PhCOOH was used instead of  $p$ -TsOH.  $^1$ 50 mol % FeCl<sub>3</sub> was used instead of  $p$ -TsOH. NP = No Product.

dichloroethane (DCE) as the solvent instead of  $CHCl<sub>3</sub>$ provided 3aa in 69% yield (Table 1, entry 3). An attempt to increase the temperature to 100 °C resulted in a lower yield, while decreasing the temperature to 65 °C or prolonging the reaction time did not give a better result (Table 1, entries 4−6). Shorting the reaction time to 3 h did not deteriorate the result and based on this, different Rh(II)-catalysts were examined (Table 1, entries 7–9). It was found that  $Rh_2(oct)_4$  was still the optimal choice, while  $Rh_2(piv)_4$  gave a lower yield and  $Rh_2(oAc)_4$  afforded a mixture. The optimal reaction conditions for 3aa were finally obtained by increasing the catalyst loading to 5 mol % and using 1.4 equiv triazole 1a (Table 1, entries 10−14). Key step for the formation of 3,6-disubstituted-1,2,4triazine 4aa, we think, is the hydrolysis of 3aa. So, 20 mol % p-TsOH as additive promoting hydrolysis was added after the rhodium-catalyzed reaction in the one-pot process. The desired product could be obtained but with an impurity difficult to separate. With this information in mind, a novel two-step method to form 3,6-disubstituted-1,2,4-triazine 4aa was set up after optimization (Table 1, entries 15−19).

After that, our next step was to explore the generality of this approach to assemble a range of 3,6-disubstituted-1,2,4-triazines under the established optimal reaction conditions. As indicated in Table 2, various substituted N-benzoylhydrazones 2a-2g reacted smoothly with the triazole 1a in moderate to excellent yie[lds to](#page-3-0) afford 3aa-3ag, while 4-pyridinyl substituted hydrazone 2h did not give the target 3ah (Table 2, entries 1−8). It should be noted that N-benzoylhydrazones 2b-2c with substituents at the ortho-position gave better [results \(T](#page-3-0)able 2, entries 2−3). Further transformations of 3 to 3,6-disubstituted-1,2,4-triazines 4 were completed very well under the [standard](#page-3-0) conditions, except for that one with strong electron-withdrawing group (Table 2, entry 5). Reactions with methyl- and benzyl- substituted N-acylhydrazones 2i-2j also successfully provided the int[ermedia](#page-3-0)tes 3ai-3aj, but the methyl-substituted one was rather stable to be hydrolyzed (Table 2, entries 9−10). After testing the limitation of N-acylhydrazones 2, we moved forward to examine varying substituti[on patte](#page-3-0)rns for various triazoles 1. As shown, the results revealed that C4 arylsubstituted triazoles 1b-1k afforded corresponding products in both rhodium-catalyzed and hydrolysis steps in good yields, which were insignificantly affected by positions or electronic properties of the substituents (Table 2, entries  $11-20$ ).<sup>14</sup> The reactions still proceeded smoothly affording products 4la and 4ma in good yields when C4 h[eteroaryl](#page-3-0)-substituted tria[zol](#page-11-0)es 1l and 1m were used (Table 2, entries 21−22).

Meanwhile, we wondered that base would promote the intramolecular cycli[zation re](#page-3-0)action by deprotonation of 3. To test our hypothesis, 1 equiv DBU was added after the rhodiumcatalyzed reaction. As expected, 3,5,6-trisubstituted-1,2,4 triazines 5aa (X-ray) was obtained in 24% yield along with unexpected 6aa in 10% yield after silic gel column chromatographic separation.<sup>15</sup> Decreasing the reaction temperature to ambient temperature, the reaction provided nearly the same result but with a l[on](#page-11-0)ger reaction time (Table 3, entries 1−2). Two equivalent DBU would obviously deteriorate the reaction and made 5aa degraded (Table 3, entr[y 3\). Afte](#page-3-0)r screening of different bases, we found that inorganic bases could not catalyze the cyclization reaction si[nce the b](#page-3-0)locking of byproduct formed in rhodium-catalyzed step in one-pot manner. Moreover, bases like  $Cs_2CO_3$ , KOH, NaOH and LDA were also not compatible with 5aa and made it degraded, while  $K_2CO_3$  catalyzed the cyclization reaction smoothly in two-step process affording 1,4,5,6-tetrahydro-1,2,4-triazines 7 in good yields and diastereoselectivities with trace amounts of 5 (Scheme 2). Also, we



<span id="page-3-0"></span>

"Standard conditions without changes. <sup>b</sup>Reaction time of step one. <sup>c</sup>Isolated yields based on substrates 2. <sup>d</sup>Isolated yields based on intermediates 3.<br>"No product substrate 1a degraded <sup>f</sup>Intermediate 3ai was stable u No product, substrate 1a degraded. Intermediate 3ai was stable under standard conditions. <sup>g</sup>A mixture was obtained when raising the temperature to 75 °C or 50 mol % FeCl<sub>3</sub> was used as additive at 50 °C.

Table 3. Optimization of Reaction Conditions for the Synthesis of 3,5,6-Trisubstituted-1,2,4-triazines  $5<sup>a</sup>$ 

Ph	$N = N$ N~Ts Ph 1) Rh(II), $T_1$ , $t_1$ 1a solvent <b>COOEt</b> 2a	Ph. 2) additive $N-Ts$ 3) [O] Ph <sup>-</sup> <b>COOEt</b> $T_2, t_2$ Заа	Ph Ph 'Ph Ph $\ddot{}$ <b>COOEt</b> 5aa 6aa	COOEt
entry	solvent	base	$T_2({}^{\circ}C)/t_2(h)$	$5$ aa/6aa $(\%)^b$
1	CHCl <sub>3</sub>	DBU/1eq	75/0.2	24/10
$\overline{2}$	CHCl <sub>3</sub>	DBU/1eq	rt/16	25/11
3	CHCl <sub>3</sub>	DBU/2eq	rt/22	$-10$
4 <sup>c</sup>	CHCl <sub>3</sub>	Et <sub>3</sub> N/3eq	rt/144	46 / 5
$5^d$	CHCl <sub>3</sub>	$DBU/0.2$ eq	30/36	45 / 5
$6^{d,e}$	CHCl <sub>3</sub>	$DBU/0.2$ eq	30/36	43 / 5
$7^{d}$ f <sub>8</sub>	DCE	$DBU/0.2$ eq	30/36	48 / 5

<sup>a</sup>Conditions: Under N<sub>2</sub>, 1a (0.24 mmol), 2a (0.2 mmol),  $Rh_2(oct)_4$  (2 mol %), and solvent (1.0 mL) were heated at 75 °C until 2a was consumed, then the next step was conducted.  $b_{\text{Isolated yields.}}$  (1 equiv Et<sub>3</sub>N was added every 48 h for three times. No additional oxidant was added.  $d_{\text{After 2a}}$ consumed, the solution was cooled to 30 °C, then 0.2 equiv DBU was added. After stirring 12 h for cyclization, 2 equiv DDQ was added for oxidation, 24 h. <sup>e</sup>1 equiv PIDA was used instead of DDQ. <sup> $f$ </sup>S mol %  $Rh_2(oct)_4$  was used. <sup>8</sup>1.4 equiv 1a was used.

found that catalytic amount of DBU was enough to catalyze the cyclization step and stoichiometric  $Et<sub>3</sub>N$  were equally effective, albeit with a longer reaction time. It is noteworthy that the oxidation process of amine in 7 to imine is a key step for

producing  $\boldsymbol{S}^{16}$  So, additional oxidants were screened for the improvement of synthesis of 3,5,6-trisubstituted-1,2,4-triazines 5aa, and 2 e[qu](#page-11-0)iv DDQ was optimal for the oxidation process (Table 1, entry 5).



<span id="page-4-0"></span>Table 4. Synthesis of 3,5,6-Trisubstituted-1,2,4-triazines  $5^a$ 



 ${}^a$ Standard conditions without changes.  ${}^b$ 0.3 equiv DBU was used as the base.  ${}^c$ 1 equiv Et<sub>3</sub>N was used as the base.





Next, attention was paid to test the scope of this protocol with various triazoles 1 and hydrazones 2, and the results are presented in Table 4. All examined substrates were effectively transformed to the desired products 5 in moderate to good yields. However, methyl-substituted N-acylhydrazone 2i and benzyl-substituted N-acylhydrazone 2j were not suitable

substrates for this transformation, as only trace amounts of the desired products were observed. However, full-substituted 1,2,4-triazoles 6ai and 6aj were successfully obtained in acceptable yields (Scheme 3). Notably, when C4 arylsubstituted triazoles with electron-donating groups or heteroarylsubstituted triazoles were used, stoichiometric  $Et<sub>3</sub>N$  was a



Scheme 5. Proposed Reaction Pathway



Scheme 6. Further Transformation of 5aa



better choice to promote cyclization step for preparation of 5ga, 5ia, 5ka, 5la, and 5ma. And also, no additive was needed in the reaction of tricyclic-triazole 1n with hydrazones 2a and 2b, and 3-substituted-naphtho $[2,1-e][1,2,4]$ triazines 8na and 8nb were obtained in good yields (Scheme 4). This method is, to our knowledge, the first case to access the 5-acyl-substituted-1,2,4-triazines directly. It should be mentioned that diaryl or hetaryl ketone frameworks are of great important that can be found in several currently marketed drugs or compounds under clinical evaluation. $17$ 

On the basis of the known N-sulfonyl-1,2,3-triazole chemistry and the [ab](#page-11-0)ove outcomes, a plausible mechanism for this divergent transformation is delineated in Scheme 5. The azavinyl carbene first reacts with the hydroxy tautomer of Nacylhydrazone 2a via an initial O−H insertion to form the putative intermediate I and releases the Rh(II)-catalyst, which will enter the next catalytic cycle to initiate the reaction. Then, intermediate I undergoes intramolecular rearrangement to afford intermediate 3aa. 3aa as the key intermediate can

undergo different routes to yield diversified products: (a) after a hydrolysis process, 3aa can transform into intermediate II. Then intramolecular condensation/aromatization processes take place successively to yield 3,6-disubstituted-1,2,4-triazine 4aa; (b) under basic conditions, 1,4,5,6-tetrahydro-1,2,4 triazine 7aa is first obtained through the intramolecular cyclization reaction, then with the help of additional oxidant or air, the desired 3,5,6-trisubstituted-1,2,4-triazine 5aa is afforded through oxidehydrogenation/aromatization steps; (c) intermediate 3aa partly degrades in the presence of base followed by intramolecular cyclization reaction to give intermediate III, which gives rise to 1,2,4-triazole 6aa after oxidation.

Finally, to demonstrate the further applications of the 1,2,4 triazines for preparing useful heterocyclic skeletons, condensation reaction and inverse electron demand Diels−Alder reaction of 5aa were examined respectively (Scheme 6). As a result, reaction of 1,2,4-triazines 5aa with hydrazine hydrate at 85 °C in ethanol was performed smoothly to afford desired

product 9aa in nearly quantitative yield, while the reaction with 2,5-norbornadiene in toluene gave ethyl 2-benzoyl-6-phenylnicotinate 10aa with  $\alpha$ -aryl pyridyl ketone moiety in 88% yield at 120 °C for 4 h. Notably,  $\alpha$ -aryl pyridyl ketone derivatives are useful intermediates in the syntheses of various natural products and drugs, which have drawn wide attention.<sup>18</sup> This method here provides a new perspective for their preparation.

# ■ CONCLUSION

In summary, we have described an unexpected O−H insertion of  $\alpha$ -imino carbenes derived from readily accessible N-sulfonyl-1,2,3-triazoles with N-acylhydrazones. Followed by intramolecular rearrangement/acid or basic-catalyzed intramolecular cyclization and oxidation reaction, it provides a facile protocol for the divergent synthesis of 3,6-disubstituted- and 3,5,6 trisubstituted-1,2,4-triazines with excellent chemoselectivities in moderate to good yields under mild conditions. To the best of our knowledge, it constitutes the first example for the synthesis of 1,2,4-triazines with different substituted-patterns from a common intermediate, which greatly enriches compound libraries of 1,2,4-triazine scaffolds with potential medicinal utilities, and the first example for the construction of multiheteroatom heteroaromatic rings using N-sulfonyl-1,2,3 triazoles as  $aze-[2C]$  synthons. Importantly, this research describes an unprecedented application of N-acylhydrazones as aze-[3C] or [4C] synthons differing from the traditional to construct azaheterocycles, which might be of potential use for the synthesis of other useful skeletons. Moreover, transformations of the 3,5,6-trisubstituted-1,2,4-triazine products were demonstrated by hydrazine condensation reaction and inverse electron demand Diels−Alder reaction, providing structurally important azaheterocycles such as pyridine or pyridazinone-fused triazine in excellent yields. Further studies on the construction of other pharmaceutically important heterocycles and biological assessment are in progress.

# **EXPERIMENTAL SECTION**

General Information. All reactions were carried out using standard Schlenk techniques under nitrogen atmosphere unless otherwise stated. CHCl<sub>3</sub>, 1,2-dichloroethane(DCE) were dried with  $CaH<sub>2</sub>$ . THF and Toluene were dried with sodium (Na). Reactions were monitored by thin layer chromatography (TLC) carried out on 0.20−0.3 mm silica gel plates (GF254) using UV light as the visualizing agent. Silica gel (200−300 mesh) was used for column chromatography. NMR spectra were recorded on 400 M instrument and calibrated using residual undeuterated solvent as an internal reference (CHCl<sub>3</sub> @ 7.26 ppm <sup>1</sup>H NMR, 77.16 ppm <sup>13</sup>C NMR). The following abbreviations (or combinations thereof) were used to explain the multiplicities:  $s = singlet$ ,  $d = doublet$ ,  $t = triplet$ ,  $q =$ quartet, m = multiplet. High-resolution mass spectra (HRMS) were recorded on Q-TOF LC/MS system. Melting points were obtained in open capillary tubes using a micro melting point apparatus which were uncorrected.  $Rh(II)$  acetate,  $Rh(II)$  octanoate were purchased from Adamas-beta.  $Rh_2(piv)_4$  were prepared using literature procedures.<sup>19</sup> N-sulfonyl-1,2,3-triazoles<sup>20−22</sup> and N-acylhydrazones were prepared according to the literature procedures.<sup>23</sup>

General Procedure [for S](#page-11-0)ynthesis of 3,6-Disubstituted-1,2,[4](#page-11-0) triazines 4. To an oven-dried Schle[nk tu](#page-11-0)be was added 0.28 mmol (1.4 equiv) N-sulfonyl-1,2,3-triazoles, 0.20 mmol (1.0 equiv) Nacylhydrazones and 0.01 mmol (5 mol %) Rh(II), successively. The Schlenk tube was sealed with a rubber plug, and the atmosphere was replaced using standard Schlenk techniques under nitrogen atmosphere. Then 1 mL of dried solvent was added and the reaction mixture was heated at 75 °C, with vigorous stirring, for 1−3 h. Once hydrazones were consumed, the reaction mixture was cooled to

ambient temperature. Then reaction mixture was purified by flash chromatography (petroleum ether/ethyl acetate = 5:1−3:1) to provide the intermediate. The intermediate was then dissolved in 2 mL of dried THF and 20 mol % p-TsOH was added. The resulting solution was stirred at 30 °C for 4 h. After reaction, solvent was evaporated and the residue was purified by flash chromatography (petroleum ether/ ethyl acetate = 10:1−5:1) to afford the desired product.

3,6-Diphenyl-1,2,4-triazine (4aa). Yellow solid, 24 mg, 51% yield, mp: 148–150 °C (literature: 158–160 °C);<sup>26</sup> <sup>1</sup>H NMR (400 MHz, chloroform-d) δ 9.06 (s, 1H), 8.65−8.53 (m, 2H), 8.17−8.15 (m, 2H), 7.63−7.51 (m, 6H). 13C NMR (100 MHz, [c](#page-11-0)hloroform-d) δ 162.5, 155.2 146.6, 134.7, 133.4, 131.8, 131.0, 129.5, 129.1, 128.3, 126.8; IR (KBr):  $\tilde{v}$  = 3049, 2957, 2923, 2853, 1964, 1905, 1655, 1636, 1599, 1452, 1405, 1320, 1178, 1087, 1025, 764, 583 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{15}H_{11}N_3$  [M]<sup>+</sup>, 233.0953; found, 233.0950.

6-Phenyl-3-(o-tolyl)-1,2,4-triazine (4ab). Yellow solid, 35 mg, 70% yield, mp: 125−127 °C (literature: 118−120 °C);<sup>27</sup> <sup>1</sup>H NMR (400 MHz, chloroform-d) δ 9.11 (s, 1H), 8.20−8.18 (m, 2H), 8.03 (d, J = 7.5 Hz, 1H), 7.62−7.58 (m, 3H), 7.46−7.42 (m, 1[H\)](#page-11-0), 7.38 (t, J = 7.2 Hz, 2H), 2.68 (s, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d)  $\delta$  165.4, 154.4, 146.1, 138.3, 134.8, 133.3, 131.8, 131.0, 130.9, 130.5, 129.5, 126.9, 126.3, 21.7; IR (KBr):  $\tilde{v}$  = 3053, 2959, 2923, 2853, 1968, 1915, 1600, 1578, 1495, 1439, 1401, 1329, 1127, 1080, 988, 932, 821, 767, 725, 694, 590  $\text{cm}^{-1}$ ; HRMS (EI-TOF): calcd for,  $\text{C}_{16}\text{H}_{13}\text{N}_3$   $[\text{M}]^+$ , 247.1109; found, 247.1110.

3-(2-Fluorophenyl)-6-phenyl-1,2,4-triazine (4ac). Yellow solid, 31 mg, 62% yield, mp: 118−120 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d) δ 9.13 (s, 1H), 8.28−8.23 (m, 1H), 8.20−8.17 (m, 2H), 7.62−7.52 (m, 4H), 7.35 (t, J = 7.6 Hz, 1H), 7.30−7.26 (m, 1H); 13C NMR (100 MHz, chloroform-d)  $\delta$  161.9 (d, J = 5.4 Hz), 161.6 (d, J = 257.3 Hz), 154.9, 146.3, 133.1 (d, J = 23.3 Hz), 133.0, 131.8, 131.2, 129.5, 126.9, 124.6 (d, J = 3.9 Hz), 123.6 (d, J = 9.6 Hz), 117.2 (d, J = 22.1 Hz); IR (KBr):  $\tilde{v}$  = 3059, 2924, 2853, 1974, 1914, 1611, 1580, 1495, 1452, 1407, 1322, 1233, 1115, 1081, 1040, 831, 769, 696 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{15}H_{10}N_3F$  [M]<sup>+</sup>, 251.0859; found, 251.0860.

6-Phenyl-3-(m-tolyl)-1,2,4-triazine (4ad). Yellow solid, 20 mg, 41% yield, mp: 108−110 °C; <sup>1</sup>H NMR (400 MHz, chloroform-*d*)  $\delta$  9.05 (s, 1H), 8.42 (s, 1H), 8.38 (d, J = 7.8 Hz, 1H), 8.18−8.16 (m, 2H), 7.60− 7.57 (m, 3H), 7.45 (t, J = 7.6 Hz, 1H), 7.38 (d, J = 7.5 Hz, 1H), 2.49 (s, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d)  $\delta$  162.6, 155.1, 146.5, 138.8, 134.7, 133.4, 132.6, 130.9, 129.5, 128.9, 128.8, 126.8, 125.5, 21.7; IR (KBr):  $\tilde{v}$  = 3046, 2922, 2852, 1961, 1898, 1816, 1598, 1493, 1431, 1395, 1331, 1090, 1039, 922, 781, 693 cm<sup>−</sup><sup>1</sup> ; HRMS (EI-TOF): calcd for,  $C_{16}H_{13}N_3$  [M]<sup>+</sup>, 247.1109; found, 247.1108.

6-Phenyl-3-(3-(trifluoromethyl)phenyl)-1,2,4-triazine (4ae). Yellow solid, 10 mg, 16% yield, mp: 156−158 °C; <sup>1</sup>H NMR (400 MHz, chloroform-d)  $\delta$  9.11 (s, 1H), 8.89 (s, 1H), 8.80 (d, J = 7.9 Hz, 1H), 8.19−8.17 (m, 2H), 7.82 (d, J = 7.7 Hz, 1H), 7.70 (t, J = 7.8 Hz, 1H), 7.63−7.60 (m, 3H); 13C NMR (100 MHz, Chloroform-d) δ 161.4, 155.8, 146.7, 135.6, 133.1, 131.6 (q, J = 32.6 Hz), 131.4, 131.4, 129.6, 129.6, 128.2 (q,  $J = 3.7$  Hz), 126.9, 125.2 (q,  $J = 3.8$  Hz), 124.1 (q,  $J =$ 272.3 Hz); IR (KBr):  $\tilde{v}$  = 3059, 2924, 2853, 1654, 1613, 1579, 1441, 1396, 1346, 1315, 1271, 1168, 1112, 1068, 923, 803, 693 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{16}H_{10}N_3F_3$  [M]<sup>+</sup>, 301.0827; found, 301.0828.

6-Phenyl-3-(p-tolyl)-1,2,4-triazine (4af). Yellow solid, 29 mg, 58% yield, mp: 161−163 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d) δ 9.04 (s, 1H), 8.48 (d, J = 8.1 Hz, 2H), 8.17−8.15 (m, 2H), 7.60−7.57 (m, 3H), 7.37 (d, J = 8.0 Hz, 2H), 2.46 (s, 3H).<sup>28</sup> <sup>13</sup>C NMR (100 MHz, chloroform-d) δ 162.6, 154.9, 146.5, 142.3, 133.5, 132.1, 130.9, 129.8, 129.5, 128.3, 126.7[, 2](#page-11-0)1.7; IR (KBr):  $\tilde{v}$  = 3031, 2920, 2853, 1926, 1739, 1654, 1607, 1507, 1442, 1404, 1331, 1181, 1086, 1033, 795, 687, 581 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for, C<sub>16</sub>H<sub>13</sub>N<sub>3</sub> [M]<sup>+</sup>, 247.1109; found, 247.1108.

3-(4-Fluorophenyl)-6-phenyl-1,2,4-triazine (4ag). Yellow solid, 21 mg, 40% yield, mp: 176−178 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d) δ 9.05 (s, 1H), 8.62−8.59 (m, 2H), 8.17−8.15 (m, 2H), 7.62−7.58 (m, 3H), 7.27–7.22 (m, 2H). <sup>13</sup>C NMR (100 MHz, chloroform-*d*) δ 165.4  $(d, J = 252.3 \text{ Hz})$ , 161.7, 155.1, 146.6, 133.3, 131.1, 130.9  $(d, J = 3.1)$ Hz), 130.5 (d,  $J = 8.8$  Hz), 129.5, 126.8, 116.2 (d,  $J = 21.9$  Hz); IR (KBr):  $\tilde{v}$  = 3054, 2923, 2853, 1972, 1912, 1600, 1509, 1442, 1407,

1227, 1155, 1084, 1040, 846, 806, 762, 701, 586 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{15}H_{10}N_3F$  [M]<sup>+</sup>, 251.0859; found, 251.0858.

3-Benzyl-6-phenyl-1,2,4-triazine (4aj). Yellow solid, 15 mg, 30% yield, mp: 143−145 °C;  $^1\rm H$  NMR (400 MHz, chloroform-*d*)  $\delta$  8.93 (s, 1H), 8.10−7.98 (m, 2H), 7.59−7.50 (m, 3H), 7.43 (d, J = 7.4 Hz, 2H), 7.33 (t, J = 7.5 Hz, 2H), 7.25 (t, J = 7.3 Hz, 1H), 4.51 (s, 2H); <sup>13</sup>C NMR (100 MHz, chloroform-d)  $\delta$  167.5, 155.4, 147.0, 137.3, 133.3, 130.9, 129.5, 129.4, 128.9, 127.1, 126.9, 43.5; IR (KBr):  $\tilde{v}$  = 3058, 2925, 2853, 1720, 1636, 1562, 1495, 1437, 1406, 1318, 1080, 1047, 1004, 919, 758, 694 cm<sup>−</sup><sup>1</sup> ; HRMS (EI-TOF): calcd for,  $C_{16}H_{13}N_3$  [M]<sup>+</sup>, 247.1109; found, 247.1110.

6-(4-Bromophenyl)-3-phenyl-1,2,4-triazine (4ba). Yellow solid, 32 mg, 51% yield, mp: 144–146 °C (literature: 201 °C);<sup>28 1</sup>H NMR (400 MHz, chloroform-d) δ 9.05 (s, 1H), 8.59−8.57 (m, 2H), 8.05 (d, J = 8.5 Hz, 2H), 7.73 (d, J = 8.4 Hz, 2H), 7.58−7.56 ([m, 3](#page-11-0)H); 13C NMR (100 MHz, chloroform-d) δ 162.7, 154.3, 146.3, 134.6, 132.8, 132.3, 131.9, 129.1, 128.4, 128.2, 125.9; IR (KBr):  $\tilde{v} = 3050, 2924, 2853,$ 1737, 1654, 1636, 1561, 1459, 1404, 1380, 1309, 1165, 1091, 617 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{15}H_{10}N_3Br^{79}$  [M]<sup>+</sup>, 311.0058; found, 311.0060.

6-(4-Chlorophenyl)-3-phenyl-1,2,4-triazine (4ca). Yellow solid, 25 mg, 47% yield, mp: 148–151 °C (literature: 197–198 °C);<sup>26 1</sup>H NMR (400 MHz, chloroform-d) δ 9.04 (s, 1H), 8.59−8.57 (m, 2H), 8.13− 8.10 (m, 2H), 7.58−7.55 (m, 5H); 13C NMR (100 MHz, [ch](#page-11-0)loroformd) δ 162.7, 154.2, 146.3, 137.5, 134.6, 131.9, 131.8, 129.8, 129.1, 128.3, 127.9; IR (KBr):  $\tilde{v}$  = 3056, 2922, 1920, 1673, 1596, 1489, 1448, 1412, 1326, 1236, 1089, 1009, 841, 749, 692, 581 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{15}H_{10}N_3Cl^{35}$  [M]<sup>+</sup>, 267.0563; found, 267.0565.

6-(4-Fluorophenyl)-3-phenyl-1,2,4-triazine (4da). Yellow solid, 21 mg, 42% yield, mp: 137−139 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d) δ 9.04 (s, 1H), 8.59−8.57 (m, 2H), 8.19−8.16 (m, 2H), 7.58−7.56 (m, 3H), 7.30−7.26 (m, 2H); <sup>13</sup>C NMR (100 MHz, chloroform-*d*) δ 164.8  $(d, J = 251.8 \text{ Hz})$ , 162.5, 154.3, 146.3, 134.7, 131.9, 129.6  $(d, J = 3.3 \text{ Hz})$ Hz), 129.1, 128.8 (d,  $J = 8.6$  Hz), 128.3, 116.7 (d,  $J = 22.0$  Hz); IR (KBr): ṽ= 3066, 2922, 2852, 1738, 1703, 1601, 1512, 1451, 1414, 1327, 1239, 1161, 1106, 1033, 842, 753, 689 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{15}H_{10}N_3F$  [M]<sup>+</sup>, 251.0859; found, 251.0858.

3-Phenyl-6-(p-tolyl)-1,2,4-triazine (4ea). Yellow solid, 26 mg, 52% yield, mp: 158–160 °C (literature: 156–157 °C);<sup>26 1</sup>H NMR (400 MHz, chloroform-d) δ 9.04 (s, 1H), 8.60−8.57 (m, 2H), 8.08−8.06 (m, 2H), 7.58−7.55 (m, 3H), 7.40−7.38 (m, 2H), [2](#page-11-0).46 (s, 3H); 13C NMR (100 MHz, chloroform-d) δ 162.3, 155.1, 146.4, 141.5, 134.9, 131.7, 130.6, 130.2, 129.0, 128.2, 126.7, 21.6; IR (KBr):  $\tilde{v} = 3056$ , 2922, 2853, 1917, 1776, 1614, 1601, 1563, 1447, 1408, 1329, 1189, 1085, 1036, 988, 830, 750, 690 cm<sup>−</sup><sup>1</sup> ; HRMS (EI-TOF): calcd for,  $C_{16}H_{13}N_3$  [M]<sup>+</sup>, 247.1109; found, 247.1110.

6-(4-methoxyphenyl)-3-phenyl-1,2,4-triazine (4fa). Yellow solid, 25 mg, 47% yield, mp: 165−167 °C (literature: 161−164 °C);<sup>27 1</sup>H NMR (400 MHz, chloroform-d) δ 9.02 (s, 1H), 8.58−8.55 (m, 2H), 8.13 (d, J = 8.7 Hz[, 2](#page-11-0)H), 7.56–7.55 (m, 3H), 7.09 (d, J = 8.7 Hz, 2H), 3.90 (s, 3H); 13C NMR (100 MHz, chloroform-d) δ 162.1, 161.9, 154.7, 146.0, 134.9, 131.6, 129.0, 128.2, 128.1, 125.7, 114.9, 55.6; IR (KBr):  $\tilde{v}$  = 3043, 2923, 2852, 1971, 1923, 1655, 1605, 1512, 1407, 1296, 1255, 1179, 1040, 988, 840, 755, 692 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{16}H_{13}N_3O$  [M]<sup>+</sup>, 263.1059; found, 263.1061.

6-(3-Methoxyphenyl)-3-phenyl-1,2,4-triazine (4ga). Yellow solid, 24 mg, 46% yield, mp: 128–130 °C; <sup>1</sup>H NMR (400 MHz, chloroformd)  $\delta$  9.06 (s, 1H), 8.60–8.58 (m, 2H), 7.82–7.81 (m, 1H), 7.68–7.65  $(m, 1H)$ , 7.59–7.55  $(m, 3H)$ , 7.49  $(t, J = 8.0 \text{ Hz}, 1H)$ , 7.11  $(ddd, J =$ 8.3, 2.6, 0.8 Hz, 1H), 3.93 (s, 3H); <sup>13</sup>C NMR (100 MHz, chloroformd) δ 162.6, 160.6, 154.9, 146.7, 134.7, 134.7, 131.8, 130.5, 129.1, 128.3, 118.9, 117.4, 111.6, 55.6; IR (KBr):  $\tilde{v}$  = 3065, 2923, 2852, 1604, 1582, 1494, 1457, 1405, 1328, 1253, 1205,1085, 1047, 891, 780, 687 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{16}H_{13}N_3O$  [M]<sup>+</sup>, 263.1059; found, 263.1058.

6-(3-Bromophenyl)-3-phenyl-1,2,4-triazine (4ha). Yellow solid, 24 mg, 38% yield, mp: 123−125 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d)  $\delta$  9.04 (s, 1H), 8.60–8.58 (m, 2H), 8.35 (s, 1H), 8.09 (d, J = 7.8 Hz, 1H), 7.70 (d, J = 8.0 Hz, 1H), 7.58−7.55 (m, 3H), 7.47 (t, J = 7.9 Hz, 1H); 13C NMR (100 MHz, chloroform-d) δ 162.9, 153.9, 146.5, 135.4,

134.5, 133.9, 132.0, 130.9, 129.8, 129.1, 128.4, 125.2, 123.7; IR (KBr): ṽ= 3064, 2954, 2923, 2852, 1949, 1634, 1564, 1446, 1407, 1319, 1089, 904, 788, 687 cm $^{-1}$ ; HRMS (EI-TOF): calcd for,  $\rm{C_{15}H_{10}N_3Br^{79}}$   $\rm{[M]^+}$ , 311.0058; found, 311.0060.

6-(2-Methoxyphenyl)-3-phenyl-1,2,4-triazine (4ia). Yellow solid, 26 mg, 49% yield, mp: 159−161 °C; <sup>1</sup>H NMR (400 MHz, chloroformd)  $\delta$  9.26 (s, 1H), 8.60–8.57 (m, 2H), 8.13 (dd, J = 7.7, 1.7 Hz, 1H), 7.58−7.55 (m, 3H), 7.51 (ddd, J = 8.4, 7.5, 1.8 Hz, 1H), 7.18 (td, J = 7.6, 0.9 Hz, 1H), 7.07 (d, J = 8.3 Hz, 1H), 3.93 (s, 3H); 13C NMR (100 MHz, chloroform-d) δ 161.6, 157.5, 155.1, 150.5, 135.1, 132.3, 131.6, 130.9, 128.9, 128.2, 122.8, 121.7, 111.5, 55.8; IR (KBr):  $\tilde{v}$  = 3057, 3004, 2924, 2853, 1753, 1658, 1601, 1494, 1457, 1406, 1270, 1179, 1088, 1020, 743, 685 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{16}H_{13}N_3O$  [M]<sup>+</sup>, 263.1059; found, 263.1060.

6-(2-Chlorophenyl)-3-phenyl-1,2,4-triazine (4ja). Yellow solid, 24 mg, 45% yield, mp: 140−142 °C (literature: 89−90 °C);<sup>26 1</sup>H NMR (400 MHz, chloroform-d) δ 9.10 (s, 1H), 8.63−8.60 (m, 2H), 7.88− 7.86 (m, 1H), 7.59−7.55 (m, 4H), 7.52−7.48 (m, 2H[\);](#page-11-0) 13C NMR (100 MHz, chloroform-d) δ 162.3, 155.9, 150.0, 134.6, 133.2, 132.6, 132.0, 131.9, 131.6, 130.5, 129.1, 128.5, 127.8; IR (KBr):  $\tilde{v} = 3095$ , 3059, 2923, 2853, 1739, 1654, 1560, 1452, 1402, 1164, 1085, 1057, 1031, 765, 687 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for, C<sub>15</sub>H<sub>10</sub>N<sub>3</sub>Cl<sup>35</sup>  $[M]$ <sup>+</sup>, 267.0563; found, 267.0568.

3-Phenyl-6-(3,4,5-trimethoxyphenyl)-1,2,4-triazine (4ka). Yellow solid, 29 mg, 45% yield, mp: 129−131 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d) δ 9.03 (s, 1H), 8.59−8.57 (m, 2H), 7.57−7.54 (m, 3H), 7.41 (s, 2H), 3.99 (s, 6H), 3.95 (s, 3H); 13C NMR (100 MHz, chloroform-d) δ 162.3, 154.6, 154.1, 146.3, 140.8, 134.6, 131.8, 129.0, 128.5, 128.2, 103.9, 61.1, 56.5; IR (KBr):  $\tilde{v} = 3051$ , 2992, 2926, 2852, 1654, 1589, 1507, 1458, 1399, 1350, 1132, 995, 882, 751, 690 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{18}H_{17}N_3O_3$  [M]<sup>+</sup>, 323.1270; found, 323.1271.

3-Phenyl-6-(thiophen-3-yl)-1,2,4-triazine (4la). Yellow solid, 15 mg, 31% yield, mp: 150−152 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d)  $\delta$  8.98 (s, 1H), 8.57–8.55 (m, 2H), 8.18–8.17 (m, 1H), 7.88 (d, J = 4.9 Hz, 1H), 7.57−7.55 (m, 4H); 13C NMR (100 MHz, chloroform-d) δ 162.1, 151.8, 146.4, 135.7, 134.8, 131.7, 129.1, 128.2, 127.8, 125.8, 125.6; IR (KBr):  $\tilde{v}$  = 3082, 2957, 2922, 2852, 1965, 1908, 1654, 1600, 1528, 1427, 1385, 1310, 1087, 1040, 960, 868, 799, 751, 690, 646 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for, C<sub>13</sub>H<sub>9</sub>N<sub>3</sub>S [M]<sup>+</sup>, 239.0517; found, 239.0518.

3-(3-Phenyl-1,2,4-triazin-6-yl)-1-(phenylsulfonyl)-1H-indole (4ma). Yellow solid, 43 mg, 52% yield, mp: 241−243 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d) δ 9.04 (s, 1H), 8.58−8.53 (m, 3H), 8.30 (s, 1H), 8.07 (d, J = 8.1 Hz, 1H), 7.98 (d, J = 7.7 Hz, 2H), 7.60−7.55 (m, 4H), 7.51–7.39 (m, 4H); <sup>13</sup>C NMR (100 MHz, chloroform-d)  $\delta$ 161.8, 152.2, 146.9, 137.8, 135.6, 134.8, 134.6, 131.8, 129.7, 129.0, 128.2, 127.7, 127.1, 126.1, 125.9, 124.7, 123.2, 116.8, 113.6; IR (KBr): ṽ= 3143, 3055, 2923, 2853, 1602, 1566, 1546, 1444, 1411, 1370, 1334, 1285, 1173, 1140, 1107, 1088, 1024, 999, 750, 686, 593 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{23}H_{16}N_4O_2S$  [M]<sup>+</sup>, 412.0994; found, 412.0996.

General Procedure for Synthesis of 3,5,6-Trisubstituted-1,2,4-triazines 5. To an oven-dried Schlenk tube was added 0.28 mmol (1.4 equiv) of N-sulfonyl-1,2,3-triazoles, 0.20 mmol (1.0 equiv) of N-acylhydrazones, and 0.01 mmol (5 mol %) of Rh(II), successively. The Schlenk tube was sealed with a rubber plug, and the atmosphere was replaced using standard Schlenk techniques under nitrogen atmosphere. Then 1 mL of dried solvent was added and the reaction mixture was heated at 75 °C, with vigorous stirring, for 1−3 h. Once the hydrazones were consumed, the reaction mixture was cooled to 30 °C, and 20 mol % DBU or 100 mol % TEA was added. About 12 h later, the intermediate was consumed, and then DDQ 0.4 mmol (2.0 equiv) was added. The resulting solution was stirred for a further 24 h. After reaction, the solvent was evaporated and the residue was purified by flash chromatography (petroleum ether/ethyl acetate = 10:1−5:1) to afford the desired product.

Ethyl 5-Benzoyl-3-phenyl-1,2,4-triazine-6-carboxylate (5aa). Yellow solid, 31 mg, 46% yield, mp: 119−121 °C; <sup>1</sup>H NMR (400 MHz, chloroform-d)  $\delta$  8.63 (d, J = 7.5 Hz, 2H), 7.89 (d, J = 7.8 Hz, 2H), 7.69 (t, J = 7.4 Hz, 1H), 7.63 (t, J = 7.2 Hz, 1H), 7.55 (q, J = 7.6 Hz, 4H), 4.43 (q, J = 7.1 Hz, 2H), 1.31 (t, J = 7.1 Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d) δ 190.3, 163.6, 162.6, 157.1, 146.4, 134.8, 134.2, 133.4, 133.4, 130.1, 129.6, 129.3, 129.1, 63.4, 13.9; IR (KBr):  $\tilde{v} = 3061$ , 2986, 2925, 2853, 1973, 1749, 1720, 1685, 1596, 1450, 1388, 1307, 1279, 1201, 1046, 968, 861, 779, 692 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{19}H_{15}N_3O_3$  [M]<sup>+</sup>, 333.1113; found, 333.1112.

Ethyl 5-Benzoyl-3-(o-tolyl)-1,2,4-triazine-6-carboxylate (5ab). Yellow solid, 38 mg, 55% yield, mp: 80−82 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d)  $\delta$  8.16 (d, J = 7.4 Hz, 1H), 7.89 (d, J = 7.5 Hz, 2H), 7.68 (t, J = 7.4 Hz, 1H), 7.53 (t, J = 7.7 Hz, 2H), 7.47 (t, J = 7.1 Hz, 1H), 7.37 (t,  $J = 6.9$  Hz, 2H), 4.44 (q,  $J = 7.1$  Hz, 2H), 2.68 (s, 3H), 1.32 (t, J = 7.1 Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d)  $\delta$ 190.2, 166.4, 162.6, 156.5, 145.6, 139.7, 134.8, 134.2, 133.2, 132.3, 131.9, 131.9, 130.1, 129.1, 126.5, 63.4, 22.2, 13.9; IR (KBr):  $\tilde{v} = 3066$ , 2964, 2925, 2853, 1988, 1723, 1684, 1597, 1507, 1369,, 1289, 1211, 1109, 966, 877, 776, 710, 657 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{20}H_{17}N_3O_3$  [M]<sup>+</sup>, 347.1270; found, 347.1271.

Ethyl 5-Benzoyl-3-(2-fluorophenyl)-1,2,4-triazine-6-carboxylate (5ac). Yellow solid, 33 mg, 47% yield, mp: 106−108 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d)  $\delta$  8.28 (td, J = 7.7, 1.5 Hz, 1H), 7.93 (d, J = 7.4 Hz, 2H), 7.69 (t, J = 7.5 Hz, 1H), 7.62−7.52 (m, 3H), 7.34 (t, J = 7.6 Hz, 1H), 7.29−7.24 (m, 1H), 4.44 (q, J = 7.1 Hz, 2H), 1.32 (t, J = 7.1 Hz, 3H); 13C NMR (100 MHz, chloroform-d) δ 189.9, 162.9 (d, J  $= 5.7$  Hz), 162.5, 162.0 (d,  $J = 260.7$  Hz), 156.5, 146.4, 134.9, 134.4  $(d, J = 8.8 \text{ Hz})$ , 134.1, 132.4, 130.7, 130.2, 129.1, 129.0, 124.7  $(d, J =$ 3.9 Hz), 122.4 (d, J = 9.1 Hz), 117.5 (d, J = 21.8 Hz), 63.4, 13.9; IR (KBr):  $\tilde{v}$  = 3068, 2980, 2924, 2853, 1986, 1725, 1679, 1611, 1498, 1452, 1385, 1286, 1208, 1110, 1065, 967, 772, 713, 652 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{19}H_{14}N_3O_3F$  [M]<sup>+</sup>, 351.1019; found, 351.1018.

Ethyl 5-Benzoyl-3-(m-tolyl)-1,2,4-triazine-6-carboxylate (5ad). Yellow solid, 26 mg, 37% yield, mp: 95−97 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d) δ 8.43−8.41 (m, 2H), 7.89 (d, J = 7.6 Hz, 2H), 7.69 (t, J = 7.4 Hz, 1H), 7.54 (t, J = 7.7 Hz, 2H), 7.47−7.43 (m, 2H), 4.42 (q,  $J = 7.1$  Hz, 2H), 2.45 (s, 3H), 1.30 (t,  $J = 7.1$  Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d) δ 190.3, 163.7, 162.6, 157.1, 146.2, 139.1, 134.8, 134.3, 134.2, 133.3, 130.1, 130.0, 129.2, 129.1, 126.8, 63.3, 21.6, 13.9; IR (KBr):  $\tilde{v} = 3066$ , 2995, 2924, 2854, 1713, 1682, 1598, 1511, 1450, 1378, 1287, 1218, 1122, 1035, 971, 798, 709, 645 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for, C<sub>20</sub>H<sub>17</sub>N<sub>3</sub>O<sub>3</sub> [M]<sup>+</sup>, 347.1270; found, 347.1271.

Ethyl 5-Benzoyl-3-(3-(trifluoromethyl)phenyl)-1,2,4-triazine-6 carboxylate (5ae). Yellow solid, 13 mg, 17% yield, mp:  $104-106$  $^{\circ}$ C; <sup>1</sup>H NMR (400 MHz, chloroform-*d*)  $\delta$  8.88 (s, 1H), 8.83 (d, J = 7.9 Hz, 1H), 7.88 (d, J = 7.8 Hz, 3H), 7.71 (t, J = 7.9 Hz, 2H), 7.55 (t, J = 7.8 Hz, 2H), 4.44 (q,  $J = 7.1$  Hz, 2H), 1.31 (t,  $J = 7.1$  Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d) δ 190.0, 162.5, 162.4, 157.3, 146.9, 134.9, 134.2, 134.1, 132.6, 131.9 (q, J = 33.0 Hz), 130.1, 129.9, 129.7  $(q, J = 3.5 \text{ Hz})$ , 129.2, 126.3  $(q, J = 3.9 \text{ Hz})$ , 123.8  $(d, J = 272.7 \text{ Hz})$ , 63.5, 13.9; IR (KBr):  $\tilde{v}$  = 3085, 2982, 2920, 2850, 1723, 1686, 1597, 1511, 1450, 1394, 1272, 1173, 1123, 971, 818, 697, 645 cm<sup>−</sup><sup>1</sup> ; HRMS (EI-TOF): calcd for,  $C_{20}H_{14}N_3O_3F_3$  [M]<sup>+</sup>, 401.0987; found, 401.0984.

Ethyl 5-Benzoyl-3-(p-tolyl)-1,2,4-triazine-6-carboxylate (5af). Yellow solid, 29 mg, 42% yield, mp: 130−132 °C; <sup>1</sup>H NMR (400 MHz, chloroform-d)  $\delta$  8.51 (d, J = 8.1 Hz, 2H), 7.88 (d, J = 7.9 Hz, 2H), 7.68  $(t, J = 7.4 \text{ Hz}, 1H), 7.53 (t, J = 7.7 \text{ Hz}, 2H), 7.35 (d, J = 8.1 \text{ Hz}, 2H),$ 4.42 (q, J = 7.1 Hz, 2H), 2.46 (s, 3H), 1.30 (t, J = 7.1 Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d) δ 190.4, 163.6, 162.7, 157.1, 146.1, 144.3, 134.7, 134.3, 130.7, 130.1, 130.1, 129.6, 129.9, 63.2, 21.9, 13.9; IR (KBr):  $\tilde{v}$  = 3067. 2992, 2924, 2855, 1981, 1720, 1677, 1608, 1502, 1449, 1371, 1288, 1208, 1178, 1045, 963, 814, 725, 653 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{20}H_{17}N_3O_3$  [M]<sup>+</sup>, 347.1270; found, 347.1272.

Ethyl 5-Benzoyl-3-(4-fluorophenyl)-1,2,4-triazine-6-carboxylate (**5ag**). Yellow solid, 28 mg, 40% yield, mp: 170–172 °C; <sup>1</sup>H NMR (400 MHz, chloroform-d) δ 8.67−8.63 (m, 2H), 7.88 (d, J = 7.5 Hz, 2H), 7.69 (t, J = 7.4 Hz, 1H), 7.54 (t, J = 7.7 Hz, 2H), 7.27−7.21 (m, 2H), 4.42 (q, J = 7.1 Hz, 2H), 1.31 (t, J = 7.1 Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d)  $\delta$  190.2, 166.2 (d, J = 255.3 Hz), 162.7, 162.5, 157.2, 146.2, 134.8, 134.2, 131.9 (d,  $J = 9.2$  Hz), 130.1, 129.6 (d,  $J =$ 3.0 Hz), 129.1, 116.5 (d, J = 22.0 Hz), 63.4, 13.9; IR (KBr):  $\tilde{v} = 3069$ , 2924, 2853, 1721, 1675, 1598, 1509, 1375, 1285, 1208, 1154, 1110,

1046, 965, 877, 788, 652 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{19}H_{14}N_3O_3F$  [M]<sup>+</sup>, 351.1019; found, 351.1020.

Ethyl 5-(4-Bromobenzoyl)-3-phenyl-1,2,4-triazine-6-carboxylate (5ba). Yellow solid, 33 mg, 40% yield, mp: 144−146 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d) δ 8.62−8.59 (m, 2H), 7.76 (d, J = 8.5 Hz, 2H), 7.68 (d, J = 8.5 Hz, 2H), 7.63 (t, J = 7.3 Hz, 1H), 7.56 (t, J = 7.5 Hz, 2H), 4.45 (q, J = 7.1 Hz, 2H), 1.35 (t, J = 7.1 Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d) δ 189.3, 163.5, 162.6, 156.5, 146.4, 133.4, 133.2, 133.0, 132.5, 131.4, 130.4, 129.5, 129.3, 63.4, 14.0; IR (KBr):  $\tilde{v}$ = 3096, 2989, 2936, 2867, 1720, 1682, 1584, 1507, 1375, 1285, 1211, 1126, 1070, 967, 879, 779, 703, 672 cm<sup>−</sup><sup>1</sup> ; HRMS (EI-TOF): calcd for,  $C_{19}H_{14}N_3O_3Br^{79}$  [M]<sup>+</sup>, 411.0219; found, 411.0218.

Ethyl 5-(4-Chlorobenzoyl)-3-phenyl-1,2,4-triazine-6-carboxylate (5ca). Yellow solid, 28 mg, 38% yield, mp: 148−151 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d)  $\delta$  8.61 (d, J = 7.5 Hz, 1H), 7.84 (d, J = 8.5 Hz, 1H), 7.63 (t, J = 7.3 Hz, 1H), 7.56 (t, J = 7.6 Hz, 1H), 7.52 (d, J = 8.5 Hz, 1H), 4.45 (q,  $J = 7.1$  Hz, 1H), 1.35 (t,  $J = 7.1$  Hz, 1H); <sup>13</sup>C NMR (100 MHz, chloroform-d) δ 189.1, 163.6, 162.7, 156.6, 146.5, 141.5, 133.4, 133.3, 132.6, 131.4, 129.5, 129.5, 129.3, 63.4, 14.0; IR (KBr): ṽ= 3096, 2968, 2923, 2852, 1720, 1683, 1586, 1508, 1377, 1284, 1211, 1115, 1089, 1044, 967, 779, 703, 672 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{19}H_{14}N_3O_3Cl^{35}$  [M]<sup>+</sup>, 367.0724; found, 367.0726.

Ethyl 5-(4-fluorobenzoyl)-3-phenyl-1,2,4-triazine-6-carboxylate (5da). yellow solid, 31 mg, 44% yield, mp: 137−139 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d) δ 8.63−8.60 (m, 2H), 7.96−7.92 (m, 2H), 7.65−7.61 (m, 1H), 7.56 (t, J = 7.4 Hz, 2H), 7.22 (t, J = 8.6 Hz, 2H), 4.45 (q,  $J = 7.1$  Hz, 2H), 1.34 (t,  $J = 7.1$  Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d) δ 188.8, 166.8 (d, J = 258.4 Hz), 163.5, 162.7, 156.7, 146.5, 133.4, 133.3, 132.9 (d, J = 9.7 Hz), 130.8 (d, J = 2.9 Hz), 129.5, 129.3, 116.5 (d, J = 22.3 Hz), 63.4, 14.0; IR (KBr):  $\tilde{v} = 3074$ , 2985, 2929, 1921, 1721, 1680, 1596, 1507, 1373, 1285, 1242, 1211, 1160, 971, 837, 691, 608 cm<sup>−</sup><sup>1</sup> ; HRMS (EI-TOF): calcd for,  $C_{19}H_{14}N_3O_3F$  [M]<sup>+</sup>, 351.1019; found, 351.1021.

Ethyl 5-(4-Methylbenzoyl)-3-phenyl-1,2,4-triazine-6-carboxylate (5ea). Yellow solid, 28 mg, 39% yield, mp: 158−160 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d)  $\delta$  8.62–8.61 (m, 1H), 7.78 (d, J = 7.2 Hz, 1H), 7.61−7.60 (m, 1H), 7.56−7.55 (m, 1H), 7.33 (d, J = 7.0 Hz, 1H), 4.46−4.39 (m, 1H), 2.46 (s, 1H), 1.33−1.29 (m, 1H); 13C NMR (100 MHz, chloroform-d) δ 189.9, 163.5, 162.6, 157.2, 146.5, 146.1, 133.4, 133.3, 131.8, 130.2, 129.8, 129.5, 129.2, 63.3, 22.1, 13.9; IR (KBr): ṽ= 3064, 2973, 2924, 1723, 1677, 1603, 1505, 1376, 1286, 1211, 1123, 1045, 966, 879, 773, 692 cm<sup>−</sup><sup>1</sup> ; HRMS (EI-TOF): calcd for,  $C_{20}H_{17}N_3O_3$  [M]<sup>+</sup>, 347.1270; found, 347.1271.

Ethyl 5-(4-Methoxybenzoyl)-3-phenyl-1,2,4-triazine-6-carboxylate (5fa). Yellow solid, 33 mg, 45% yield, mp: 165−167 °C; <sup>1</sup>H NMR (400 MHz, chloroform-d) δ 8.63−8.61 (m, 2H), 7.89−7.86 (m, 2H), 7.64−7.60 (m, 1H), 7.57−7.53 (m, 2H), 7.02−6.98 (m, 2H), 4.43 (q,  $J = 7.1$  Hz, 2H), 3.90 (s, 3H), 1.32 (t,  $J = 7.1$  Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d) δ 188.9, 164.9, 163.5, 162.7, 157.2, 146.7, 133.5, 133.2, 132.6, 129.5, 129.3, 127.4, 114.5, 63.2, 55.8, 13.9; IR (KBr):  $\tilde{v}$  = 3082, 2987, 2932, 2836, 1724, 1671, 1598, 1511, 1454, 1387, 1323, 1269, 1173, 1125, 1023, 968, 856, 777, 694, 610 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{20}H_{17}N_3O_4$  [M]<sup>+</sup>, 363.1219; found, 363.1218.

Ethyl 5-(3-Methoxybenzoyl)-3-phenyl-1,2,4-triazine-6-carboxylate (5ga). Yellow solid, 35 mg, 48% yield, mp: 128–130 °C; <sup>1</sup>H NMR (400 MHz, chloroform-d)  $\delta$  8.63–8.61 (m, 2H), 7.62 (t, J = 7.3 Hz, 1H), 7.57–7.54 (m, 3H), 7.40 (t, J = 7.9 Hz, 1H), 7.29 (d, J = 7.7 Hz, 1H), 7.22 (dd, J = 8.2, 2.0 Hz, 1H), 4.44 (q, J = 7.1 Hz, 2H), 3.88  $(s, 3H)$ , 1.33  $(t, J = 7.1 \text{ Hz}, 3H)$ ; <sup>13</sup>C NMR (100 MHz, chloroform-d) δ 190.1, 163.6, 162.6, 160.2, 157.1, 146.3, 135.5, 133.4, 133.3, 130.1, 129.5, 129.3, 123.4, 121.6, 113.3, 63.3, 55.7, 13.9; IR (KBr):  $\tilde{v} = 3070$ , 2989, 2958, 2922, 1720, 1683, 1596, 1512, 1466, 1385, 1285, 1194, 1109, 998, 767, 692 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{20}H_{17}N_3O_4$ [M]<sup>+</sup>, 363.1219; found, 363.1221.

Ethyl 5-(3-Bromobenzoyl)-3-phenyl-1,2,4-triazine-6-carboxylate (5ha). Yellow solid, 25 mg, 30% yield, mp: 123−125 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d)  $\delta$  8.62 (d, J = 7.4 Hz, 2H), 8.06 (s, 1H), 7.81 (d, J = 7.9 Hz, 1H), 7.77 (d, J = 7.8 Hz, 1H), 7.64 (t, J = 7.3 Hz, 1H), 7.57 (t, J = 7.5 Hz, 2H), 7.41 (t, J = 7.9 Hz, 1H), 4.46 (q, J = 7.1

Hz, 2H), 1.36 (t, J = 7.1 Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d) δ 188.9, 163.6, 162.6, 156.5, 146.3, 137.6, 135.9, 133.5, 133.2, 132.7, 130.6, 129.6, 129.3, 128.7, 123.5, 63.5, 14.0; IR (KBr):  $\tilde{v}$  = 3068, 2924, 2853, 1970, 1716, 1684, 1567, 1509, 1425, 1373, 1309, 1286, 1207, 1123, 1045, 782, 723, 670 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{19}H_{14}N_3O_3Br^{79}$  [M]<sup>+</sup>, 411.0219; found, 411.0214.

Ethyl 5-(2-Methoxybenzoyl)-3-phenyl-1,2,4-triazine-6-carboxylate (5ia). Yellow solid, 25 mg, 34% yield, mp: 159–161 °C; <sup>1</sup>H NMR (400 MHz, chloroform-d) δ 8.62−8.60 (m, 2H), 8.23 (dd, J = 7.8, 1.7 Hz, 1H), 7.65−7.57 (m, 2H), 7.55−7.51 (m, 2H), 7.19 (t, J = 8.0 Hz, 1H), 6.91 (d, J = 8.3 Hz, 1H), 4.42 (q, J = 7.1 Hz, 1H), 3.45 (s, 2H), 1.30 (t, J = 7.1 Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d)  $\delta$ 188.8, 163.7, 162.8, 160.4, 159.9, 144.9, 136.7, 133.8, 132.9, 131.2, 129.4, 129.1, 123.7, 121.7, 112.3, 63.0, 55.8, 14.0; IR (KBr):  $\tilde{v} = 3073$ , 2924, 2852, 1725, 1672, 1596, 1510, 1376, 1298, 1201, 1017, 967, 875, 760, 666, 646 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{20}H_{17}N_3O_4$   $[M]^+$ , 363.1219; found, 363.1220.

Ethyl 5-(2-Chlorobenzoyl)-3-phenyl-1,2,4-triazine-6-carboxylate (5ja). Yellow solid, 27 mg, 37% yield, mp: 140−142 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d)  $\delta$  8.56 (d, J = 7.9 Hz, 2H), 8.06 (d, J = 8.6 Hz, 1H), 7.61−7.50 (m, 4H), 7.49−7.45 (m, 2H), 4.48 (q, J = 7.1 Hz, 2H), 1.38 (t, J = 7.1 Hz, 3H);  $^{13}$ C NMR (100 MHz, chloroform-d)  $\delta$ 189.6, 163.7, 162.8, 157.7, 146.1, 135.0, 134.7, 133.4, 133.2, 133.1, 132.7, 131.4, 129.4, 129.2, 127.5, 63.4, 14.0; IR (KBr):  $\tilde{v}$  = 3066, 2921, 2851, 1992, 1721, 1672, 1584, 1508, 1376, 1293, 1202, 1061, 963, 872, 781, 746, 701, 634 cm<sup>−</sup><sup>1</sup> ; HRMS (EI-TOF): calcd for,  $C_{19}H_{14}N_3O_3Cl^{35}$  [M]<sup>+</sup>, 367.0724; found, 367.0722.

Ethyl 3-Phenyl-5-(3,4,5-trimethoxybenzoyl)-1,2,4-triazine-6-carboxylate (5ka). Yellow solid, 39 mg, 46% yield, mp: 129−<sup>131</sup> °C; <sup>1</sup> <sup>1</sup>H NMR (400 MHz, chloroform-d)  $\delta$  8.65–8.62 (m, 2H), 7.63 (t, J = 7.3 Hz, 1H), 7.57 (t, J = 7.4 Hz, 2H), 7.15 (s, 2H), 4.47 (q, J = 7.1 Hz, 2H), 3.97 (s, 3H), 3.85 (s, 6H), 1.37 (t, J = 7.1 Hz, 3H); 13C NMR (100 MHz, chloroform-d) δ 189.1, 163.4, 162.8, 156.8, 153.5, 146.8, 144.3, 133.4, 133.4, 129.4, 129.3, 129.1, 107.7, 63.3, 61.2, 56.5, 14.0; IR (KBr): ṽ= 3093, 2985, 2940, 2848, 1731, 1674, 1581, 1504, 1463, 1415, 1333, 1271, 1125, 1000, 769, 705, 687 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{22}H_{21}N_3O_6$  [M]<sup>+</sup>, 423.1430; found, 423.1431.

Ethyl 3-Phenyl-5-(thiophene-3-carbonyl)-1,2,4-triazine-6-carboxylate (5la). Yellow solid, 24 mg, 36% yield, mp: 150–152 °C; <sup>1</sup>H NMR (400 MHz, chloroform-d)  $\delta$  8.62 (dt, J = 7.1, 1.3 Hz, 2H), 8.18  $(dd, J = 2.8, 1.2 \text{ Hz}, 1H), 7.70 \text{ (dd, } J = 5.1, 1.2 \text{ Hz}, 1H), 7.66 - 7.61 \text{ (m, }$ 1H), 7.59−7.55 (m, 2H), 7.46 (dd, J = 5.1, 2.9 Hz, 1H), 4.47 (q, J = 7.1 Hz, 2H), 1.37 (t, J = 7.1 Hz, 3H); 13C NMR (100 MHz, chloroform-d) δ 183.7, 163.5, 162.9, 155.5, 147.2, 139.1, 136.9, 133.4, 133.3, 129.4, 129.3, 127.7, 127.4, 63.3, 14.0; IR (KBr):  $\tilde{v}$  = 3080, 2981, 2931, 2871, 1744, 1674, 1506, 1402, 1305, 1199, 1048, 833, 768, 702, 669 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for, C<sub>17</sub>H<sub>13</sub>N<sub>3</sub>O<sub>3</sub>S [M]<sup>+</sup>, 339.0678; found, 339.0679.

Ethyl 3-Phenyl-5-(1-(phenylsulfonyl)-1H-indole-3-carbonyl)- 1,2,4-triazine-6-carboxylate (5ma). Yellow solid, 52 mg, 51% yield, mp: 241−243 °C; <sup>1</sup>H NMR (400 MHz, chloroform-*d*) δ 8.65 (d, J = 8.5 Hz, 2H), 8.48 (s, 1H), 8.43−8.41 (m, 1H), 8.03−8.01 (m, 1H), 7.97 (d, J = 7.9 Hz, 2H), 7.69−7.61 (m, 4H), 7.53 (t, J = 7.8 Hz, 2H), 7.49−7.42 (m, 2H), 4.48 (q, J = 7.1 Hz, 2H), 1.34 (t, J = 7.1 Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d)  $\delta$  184.7, 163.5, 163.4, 153.6, 148.1, 137.4, 136.2, 135.0, 134.9, 133.5, 133.3, 129.9, 129.4, 129.3, 127.8, 127.4, 126.7, 125.7, 123.3, 118.4, 113.4, 63.3, 14.0; IR (KBr):  $\tilde{v}$ = 3127, 3065, 2959, 2924, 2854, 1974, 1723, 1660, 1540, 1446, 1394, 1369, 1300, 1265, 1176, 1083, 982, 755, 683, 593 cm<sup>−</sup><sup>1</sup> ; HRMS (EI-TOF): calcd for,  $C_{27}H_{20}N_4O_5S$  [M]<sup>+</sup>, 512.1154; found, 512.1161.

Ethyl 4-(2-Oxo-2-phenylethyl)-5-phenyl-4H-1,2,4-triazole-3-carboxylate (6aa). Yellow oil, 14 mg, 20% yield (2 equiv. NaOH as base in two-step way);  $^1\mathrm{H}$  NMR (400 MHz, chloroform-d)  $\delta$  8.52 (d, J  $= 7.7$  Hz, 2H), 8.18 (d, J = 7.2 Hz, 2H), 7.68 (t, J = 7.3 Hz, 1H), 7.55 (t, J = 7.5 Hz, 2H), 7.52−7.39 (m, 3H), 5.45 (s, 2H), 4.25 (q, J = 7.1 Hz, 2H), 1.26 (t, J = 7.1 Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d) δ 183.3, 166.9, 161.1, 150.3, 135.6, 134.4, 131.4, 130.2, 129.9, 128.8, 128.6, 126.7, 62.3, 53.1, 14.2; IR (KBr):  $\tilde{v} = 3050$ , 2924, 2853, 1752, 1654, 1561, 1459, 1404, 1282, 1209, 1092, 918, 686 cm<sup>−</sup><sup>1</sup> ; HRMS (EI-TOF): calcd for,  $C_{19}H_{17}N_3O_3$  [M]<sup>+</sup>, 335.1270; found, 335.1271.

Ethyl 5-Methyl-4-(2-oxo-2-phenylethyl)-4H-1,2,4-triazole-3-carboxylate (6ai). Yellow oill, 11 mg, 21% yield; <sup>1</sup>H NMR (400 MHz, chloroform-d)  $\delta$  8.36 (d, J = 8.0 Hz, 2H), 7.64 (t, J = 7.4 Hz, 1H), 7.52  $(t, J = 7.7 \text{ Hz}, 2H), 5.33 \text{ (s, 2H)}, 4.23 \text{ (q, } J = 7.1 \text{ Hz}, 2H), 2.50 \text{ (s, }$ 3H), 1.26 (t, J = 7.1 Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d)  $\delta$ 183.5, 167.0, 160.0, 149.9, 135.7, 134.3, 131.1, 128.6, 62.3, 52.5, 14.2, 14.1; IR (KBr):  $\tilde{v}$  = 3068, 2960, 2927, 2854, 1754, 1658, 1597, 1478, 1434, 1375, 1280, 1214, 1127, 1025, 922, 736, 692 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{14}H_{15}N_3O_3$  [M]<sup>+</sup>, 273.1113; found, 273.1111.

Ethyl 5-Benzyl-4-(2-oxo-2-phenylethyl)-4H-1,2,4-triazole-3-carboxylate (6aj). Yellow oill, 21 mg, 30% yield; <sup>1</sup>H NMR (400 MHz, chloroform-d)  $\delta$  8.37 (d, J = 7.6 Hz, 2H), 7.64 (t, J = 7.4 Hz, 1H), 7.51 (t, J = 7.7 Hz, 2H), 7.37−7.30 (m, 4H), 7.23 (d, J = 7.1 Hz, 1H), 5.34  $(s, 2H)$ , 4.22  $(q, J = 7.2$  Hz, 2H), 4.18  $(s, 2H)$ , 1.25  $(t, J = 7.1$  Hz, 3H); 13C NMR (100 MHz, chloroform-d) δ 183.4, 166.9, 162.3, 150.1, 137.7, 135.6, 134.3, 131.2, 128.9, 128.7, 128.6, 126.8, 62.3, 52.6, 34.8, 14.2; IR (KBr):  $\tilde{v}$  = 3063, 2958, 2928, 2855, 1753, 1702, 1658, 1597, 1475, 1354, 1281, 1215, 1127, 1025, 922, 875, 732, 691, 576 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{20}H_{19}N_3O_3$  [M]<sup>+</sup>, 349.1426; found, 349.1421.

General Procedure for Synthesis of 1,4,5,6-Tetrahydro-1,2,4-triazines 7. To an oven-dried Schlenk tube was added 0.28 mmol (1.4 equiv) of N-sulfonyl-1,2,3-triazoles, 0.20 mmol (1.0 equiv) of N-acylhydrazones, and 0.01 mmol (5 mol %) of Rh(II), successively. The Schlenk tube was sealed with a rubber plug, and the atmosphere was replaced using standard Schlenk techniques under nitrogen atmosphere. Then 1 mL of dried solvent was added, and the reaction mixture was heated at 75 °C, with vigorous stirring, for 1−3 h. Once the hydrazones were consumed, the reaction mixture was cooled to ambient temperature. Then reaction mixture was purified by flash chromatography (petroleum ether/ethyl acetate = 5:1−3:1) to provide the intermediate. The intermediate was then dissolved in 2 mL of dried THF, and 20 mol %  $K_2CO_3$  was added. The resulting solution was stirred at 30 °C for 12 h. After reaction, solvent was evaporated and the residue was purified by flash chromatography (petroleum ether/ethyl acetate = 10:1−5:1) to afford the desired product.

3-(3-Phenyl-1,2,4-triazin-6-yl)-1-(phenylsulfonyl)-1H-indole (7aa). Yellow solid, 60 mg, 66% yield, mp: 151−153 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d)  $\delta$  8.08 (d, J = 7.6 Hz, 2H), 7.67 (t, J = 7.4 Hz, 1H), 7.60 (d, J = 8.2 Hz, 2H), 7.52 (t, J = 7.7 Hz, 2H), 7.31 (d, J = 8.1 Hz, 2H), 7.17 (t, J = 7.2 Hz, 1H), 7.08 (t, J = 7.6 Hz, 2H), 6.99 (d,  $J = 7.5$  Hz, 2H), 6.41 (d,  $J = 3.9$  Hz, 1H), 6.03 (d,  $J = 5.2$  Hz, 1H), 4.28−4.18 (m, 2H), 3.63 (t, J = 4.7 Hz, 1H), 2.45 (s, 3H), 1.23 (t, J = 7.2 Hz, 3H); 13C NMR (100 MHz, chloroform-d) δ 192., 168.4, 145.6, 135.2, 134.7, 134.1, 134.0, 130.7, 130.4, 129.6, 128.9, 128.5, 127.8, 127.7, 126.9, 62.2, 54.7, 54.5, 21.9, 14.1; IR (KBr):  $\tilde{v} = 3372$ , 3061, 2961, 2924, 2854, 1732, 1690, 1596, 1446, 1366, 1307, 1229, 1166, 1091, 993, 767, 743, 699,666, 612 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{26}H_{25}N_3O_5S$  [M]<sup>+</sup>, 491.1515; found, 491.1513.

Ethyl 5-Benzoyl-3-(o-tolyl)-4-tosyl-1,4,5,6-tetrahydro-1,2,4-triazine-6-carboxylate (7ab). Yellow solid, 76 mg, 75% yield, mp: 166−168 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d) δ 8.06 (d, J = 7.5 Hz, 2H), 7.67 (t, J = 7.4 Hz, 1H), 7.51 (t, J = 7.7 Hz, 2H), 7.47 (d, J = 8.2 Hz, 2H), 7.21 (d,  $J = 8.1$  Hz, 2H), 7.04 (d,  $J = 3.7$  Hz, 2H), 6.68  $(dt, J = 8.3, 4.1 Hz, 1H), 6.36 (d, J = 4.0 Hz, 1H), 6.22 (d, J = 7.7 Hz,$ 1H), 6.18 (d, J = 4.8 Hz, 1H), 4.22 (q, J = 6.9 Hz, 2H), 3.88 (t, J = 4.6 Hz, 1H), 2.41 (s, 3H), 2.28 (s, 3H), 1.22 (t,  $J = 7.1$  Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d) δ 192.8, 168.2, 145.2, 137.2, 134.9, 134.8, 134.0, 133.7, 130.8, 130.0, 129.6, 129.5, 129.4, 128.9, 128.2, 127.9, 124.9, 62.2, 55.4, 53.5, 21.8, 20.6, 14.1; IR (KBr):  $\tilde{v} = 3388$ , 2962, 2922, 2852, 1741, 1688, 1595, 1453, 1332, 1301, 1231, 1158, 1086, 1036, 973, 751, 679, 616 cm<sup>−</sup><sup>1</sup> ; HRMS (EI-TOF): calcd for,  $C_{27}H_{27}N_3O_5S$  [M]<sup>+</sup>, 505.1671; found, 505.1670.

Ethyl 5-Benzoyl-3-(2-fluorophenyl)-4-tosyl-1,4,5,6-tetrahydro-1,2,4-triazine-6-carboxylate (7ac). Yellow solid, 56 mg, 55% yield, mp: 174−176 °C; <sup>1</sup>H NMR (400 MHz, chloroform-*d*) δ 8.16 (d, J = 7.8 Hz, 2H), 7.68 (t, J = 7.3 Hz, 1H), 7.63 (d, J = 8.1 Hz, 2H), 7.55 (t, J = 7.5 Hz, 2H), 7.32 (d, J = 8.0 Hz, 2H), 7.20−7.13 (m, 1H), 6.92− 6.84 (m, 2H), 6.70 (t, J = 7.6 Hz, 1H), 6.47 (d, J = 3.3 Hz, 1H), 6.05  $(d, J = 5.0$  Hz, 1H), 4.28–4.13 (m, 2H), 3.51 (t,  $J = 4.5$  Hz, 1H), 2.44

<span id="page-10-0"></span> $(s, 3H)$ , 1.20 (t, J = 7.0 Hz, 2H); <sup>13</sup>C NMR (100 MHz, chloroform-d)  $\delta$  192.2, 168.1, 160.1 (d, J = 252.1 Hz), 145.6, 134.5, 134.4, 134.1, 130.5, 130.3, 130.1 (d, J = 8.2 Hz), 129.7, 128.9, 127.8, 125.2 (d, J = 2.6 Hz), 123.7 (d,  $J = 10.8$  Hz), 123.5 (d,  $J = 3.7$  Hz), 115.9 (d,  $J =$ 21.7 Hz), 62.2, 54.5, 53.4, 21.9, 14.1; IR (KBr):  $\tilde{v}$  = 3378, 3053, 2962, 2922, 2852, 1731, 1689, 1596, 1494, 1451, 1365, 1309, 1227, 1163, 1090, 1040, 767, 666, 554 cm<sup>−</sup><sup>1</sup> ; HRMS (EI-TOF): calcd for,  $C_{26}H_{24}N_3O_5SF$  [M]<sup>+</sup>, 509.1421; found, 509.1420.

3-Phenylnaphtho[2,1-e][1,2,4]triazine (8na). Yellow solid, 20 mg, 41% yield, mp: 142−144 °C (literature: 117−118 °C)29; <sup>1</sup> H NMR (400 MHz, chloroform-d)  $\delta$  9.53 (d, J = 8.0 Hz, 1H), 8.80–8.78 (m, 2H), 8.23 (d, J = 9.1 Hz, 1H), 7.98 (d, J = 7.7 Hz, 1H), [7.9](#page-11-0)2−7.82 (m, 3H), 7.64-7.60 (m, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d) δ 161.4, 144.9, 143.2, 138.3, 135.8, 132.8, 131.5, 130.3, 129.5, 129.4, 129.1, 128.7, 128.6, 125.9, 124.1; IR (KBr):  $\tilde{v} = 3054$ , 2923, 2853, 1598, 1517, 1467, 1433, 1383, 1276, 1138, 1047, 848, 737, 690 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{17}H_{11}N_3$  [M]<sup>+</sup>, 257.0953; found, 257.0952.

3-(o-Tolyl)naphtho[2,1-e][1,2,4]triazine (8nb). Yellow solid, 25 mg, 46% yield, mp: 132−134 °C; <sup>1</sup> H NMR (400 MHz, chloroform-d)  $\delta$  9.58 (d, J = 7.9 Hz, 1H), 8.26 (d, J = 9.1 Hz, 1H), 8.13–8.11 (m, 1H), 8.01 (d, J = 7.8 Hz, 1H), 7.94−7.85 (m, 3H), 7.48−7.41 (m, 3H), 2.72 (s, 3H); 13C NMR (100 MHz, chloroform-d) δ 164.6, 144.2, 142.8, 138.4, 138.3, 135.9, 132.9, 131.7, 131.4, 130.4, 130.3, 129.5, 129.3, 128.7, 126.4, 125.8, 124.2, 21.6; IR (KBr):  $\tilde{v} = 3058$ , 2921, 2853, 1968, 1597, 1516, 1438, 1383, 1330, 1142, 1033, 851, 813, 745, 652 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for, C<sub>18</sub>H<sub>13</sub>N<sub>3</sub> [M]<sup>+</sup>, 271.1109; found, 271.1110.

General Procedure for Further Transformation of 3,5,6- Trisubstituted-1,2,4-triazine 5aa. Reaction a: with Hydrazine Hydrate. A total of 0.1 mmol (1.0 equiv) was dissolved in 2 mL of absolute ethanol. To this solution was added hydrazine hydrate, 0.15 mmol (80%, 1.5 equiv). The resulting solution was heated at 85 °C for 1 h. After reaction, the solvent was evaporated and the residue was recrystallized from hexane containing a small amount of DCM to afford the desired product 9aa, 29 mg (97% yield).

3,5-Diphenylpyridazino[4,5-e][1,2,4]triazin-8(7H)-one (9aa). Yellow solid, 29 mg, 97% yield, mp: > 250 °C; <sup>1</sup>H NMR (400 MHz, chloroform-d)  $\delta$  10.62 (s, 1H), 8.69 (d, J = 7.3 Hz, 2H), 8.05–8.02 (m, 2H), 7.68−7.65 (m, 1H), 7.62−7.59 (m, 5H); 13C NMR (125 MHz, chloroform-d) δ 165.3, 146.9, 142.8, 142.5, 133.8, 133.7, 132.1, 130.4, 129.9, 129.8, 129.8, 129.5, 128.7; IR (KBr):  $\tilde{v} = 3450, 3064,$ 2955, 2923, 2854, 1689, 1508, 1460, 1354, 1215, 1086, 881, 698 cm<sup>−</sup><sup>1</sup> ; HRMS (EI-TOF): calcd for,  $C_{17}H_{11}N_5O$  [M]<sup>+</sup>, 301.0964; found, 301.0962.

Reaction b: with 2,5-Norbornadiene. A total of 0.2 mmol (1.0 equiv) was dissolved in 2 mL of dried toluene. To this solution was added 2,5-norbornadiene, 0.2 mL (10 equiv., 2 mmol). The resulting solution was heated at 120 °C for 4 h. After reaction, the reaction mixture was purified by flash chromatography (petroleum ether/ethyl acetate =  $5:1-3:1$ ) to provide the desired product 10aa, 58 mg (88% yield).

Ethyl 2-Benzoyl-6-phenylnicotinate (10aa). Yellow solid, 58 mg, 88% yield, mp: 112−114 °C; <sup>1</sup>H NMR (400 MHz, chloroform-d)  $\delta$ 8.44 (d, J = 8.4 Hz, 1H), 8.10–8.07 (m, 2H), 7.94 (d, J = 8.4 Hz, 1H), 7.88 (d, J = 7.4 Hz, 2H), 7.59 (t, J = 7.4 Hz, 1H), 7.48−7.44 (m, 5H), 4.21 (q,  $J = 7.1$  Hz, 2H), 1.11 (t,  $J = 7.1$  Hz, 3H); <sup>13</sup>C NMR (100 MHz, chloroform-d) δ 194.2, 164.9, 159.7, 158.7, 139.3, 137.5, 135.9, 133.5, 130.4, 130.0, 129.0, 128.6, 127.7, 123.2, 120.1, 61.9, 13.7; IR (KBr):  $\tilde{v}$  = 3059, 2959, 2923, 2852, 1717, 1675, 1582, 1451, 1294, 1168, 1095, 955, 760, 707, 641 cm<sup>-1</sup>; HRMS (EI-TOF): calcd for,  $C_{21}H_{17}NO_3$  [M]<sup>+</sup>, 331.1208; found, 331.1207.

## ■ ASSOCIATED CONTENT

#### **S** Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.6b02846.

NMR (PDF) [Crystallographic dat](http://pubs.acs.org)a (CIF)

# ■ AUTHOR INFORMATION

#### Corresponding Authors

\*E-mail: xxyu@ecust.edu.cn.

\*E-mail: weiping\_deng@ecust.edu.cn

# ORCID<sup>®</sup>

Xingxin [Yu:](mailto:xxyu@ecust.edu.cn) [0000-0002-8370-2598](mailto:weiping_deng@ecust.edu.cn)

#### Notes

The authors [declare no competi](http://orcid.org/0000-0002-8370-2598)ng financial interest.

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